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Department of Chemistry, University College Cardiff, P.O. Box 78, Cardiff

Abstract

E.s.r., ENDOR, and TRIPLE resonance results are reported for the radical cation of N,N'-diphenyl-4,4'bipyridylium dichloride and a series of its fluorophenyl analogues. The effect of ortho-, meta-, and parasubstitution on the spin density distribution within the molecule is discussed and the observed coupling constants for the fluorine-substituted phenyl derivatives are analysed with the aid of INDO molecular orbital calculations. The magnitude of the spin delocalised into the phenyl rings when ortho-substituted suggests that these rings are substantially twisted with respect to the plane of the bipyridylium rings.